An *In Silico* QSAR Model Study Using Electrophilicity as a Possible Descriptor Against *T. Brucei*

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ABSTRACT

Human African trypanosomiasis (HAT) is a vector-borne sleeping sickness parasitic disease spread through the bite of infected tsetse flies (Glossina genus), which is highly populated in rural Africa. The present study constructed quantitative structure-activity relationship (QSAR) models based on quantum chemical electronic descriptors to bring out the extent to which the electronic factor of the selected compounds affects the HAT activity. Theoretical prediction of toxicity (pIC_{50}) of the series of heterocyclic scaffolds consisting 32 pyridyl benzamide derivatives towards HAT is investigated by considering all possible combinations of electrophilicity index (ω) and the square of electrophilicity index (ω^2) as descriptors in the studied models along with other descriptors previously used by Masand et al. A multiple linear regression (MLR) analysis is conducted to develop the models. Further, in order to obtain the variable selection on the overall data set having diverse functional groups, the analysis using sum of ranking differences methodology with ties is carried out.

KEYWORDS

Electrophilicity, Global Electronic Descriptor, Human African Trypanosomiasis, Multiple Linear Regression, QSAR, Sum of Ranking Differences

INTRODUCTION

Human African trypanosomiasis, commonly known as sleeping sickness is an endemic disease in the sub-Saharan African countries caused by two forms of the *Trypanosoma brucei* parasite, namely, *Trypanosoma brucei gambiense* and *Trypanosoma brucei rhodesiense* (Sykes et al., 2012;Ferrins et al., 2014). According to the World Health Organisation, 98% of the reported cases are caused by the former species resulting in a more severe form of the disease-causing chronic infection, the symptoms of which mainly remain hidden until the central nervous system is seriously affected. *Trypanosoma brucei rhodesiense*, on the other hand, accounts for the remaining 2%, causing acute infection and showing symptoms in the early stages of the disease. This vector-borne disease is known to be

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transmitted by the bite of the infected tsetse fly and is very rarely identified in its first stage where drug administration is easier and less toxic compared to those required in the most commonly identifiable second stage (Barrett et al., 2011;Simarro et al., 2012;Seixas et al., 2013;Carvalho et al., 2014).

High-throughput screening *i.e.*, Alamar Blue based, 384-well HTS assay was applied by Sykes *et al.* (2012) to screen a huge dataset of 87,296 compounds against *T. brucei* where unfortunately pyridyl benzamide derivatives were excluded by the selection criteria. Later, Ferrins *et al.* (2014) showed the HAT activity of the aforementioned class of compounds suggesting their ability to treat the disease in its second stage. However, the search for experimental targets for such studies still possesses substantial challenges. This is where computational drug design and statistical approaches like quantitative structure-activity relationships (QSARs) come in handy as computational resources are easy to work with, cheaper and faster than conducting actual experiments. QSAR provides a statistical approach in understanding the correlation between structural features of a certain class of compounds and the biological behaviour exhibited by them.

Quantitative structure-activity relationships (QSARs) are of paramount importance which pay attention to the theoretical toxicity predictions in the highly complicated field of pharmaceutical sciences due to their potential in the assessment of various biological activities (e.g., drug activity, toxicity, etc.) and physicochemical features of bioactive molecules by evading the time consuming and cost-effective experiments. It is a very interesting aspect that molecular descriptors or chemical attributes are the fingerprints of the molecular or chemical structures and thus the choice of appropriate descriptors for a particular biological property/activity/toxicity has become quite a challenging task. Cheminformatics is the initial step in such studies that efficiently utilizes computer information techniques in the field of chemistry to search, extract and process relevant information of chemical compounds from huge databases. In cheminformatics analysis, machine learning techniques are also used in QSAR studies for chemical feature extraction and selection in multiple levels, characterization of the compounds by substructure fragments as well as a selection of the chemical descriptors. These approaches have wide applications in the chemical industry in modeling physicochemical properties of chemical compounds, performing virtual screening in pharmaceutical studies, and to predict pharmacodynamic and pharmacokinetic properties in computational drug designing. The field of cheminformatics specializes in handling huge datasets, performing direct data to knowledge mapping and thus allows an easy transition from quantum chemistry to biological activity. It can provide information that may be used in medicine, biology, and physics. Various forms of machine-readable information are included in chemical databases which may be used in clinical purposes. There are some limitations of cheminformatics when used in stereoisomers and analyzing tautomers.

In this present study, we have explored the effect of the electronic atmosphere of 32 pyridyl benzamide derivatives on its HAT activity against *T. brucei*. The dataset is taken from a study performed by Masand *et al.* (2016) where they had screened a large pool of descriptors via objective and subjective feature selection and created multiple QSAR models by randomly splitting the datasets into training and prediction sets. Here we have incorporated global electrophilicity index (ω) and the square of electrophilicity index (ω^2) as electronic descriptors by replacing those used by Masand *et al.* one or two at a time to obtain all possible combinations of QSAR models. Multiple linear regression (MLR) method is employed to construct the models, followed by comparing them using the Sum of Ranking Differences (SRDs) technique described by K. Héberger (2010; Kollár-Hunek and Héberger, 2013).

THEORETICAL BACKGROUND

The electronic parameter of any drug brings out information about the polarising effect of any electronegative center present in the molecule on its binding pattern with the protein (DNA). Thus including the electronic parameters in simulating these types of bonding mechanisms is of utmost importance. Global electrophilicity index (ω) can be computed within the domain of conceptual

density functional theory (CDFT) (Hohenberg and Kohn, 1964;Kohn and Sham, 1965) using quantum chemical properties of a molecule like chemical hardness (η) (Chattaraj and Parr, 1993; Pearson, 1997) and chemical potential (μ) employing Parr's definition (Parr and Yang, 1989) as:

$$\omega = \frac{\mu^2}{2\eta} \tag{1}$$

Parr also established a quantum chemical relation between chemical potential (μ) and electronegativity (χ) (Pauling, 1960; Sen and Jorgenson, 1987) using Iczkowski and Margrave's (1961) definition of χ as the negative of first derivative of energy (E) with respect to the number of electrons (N) at the constant external potential $v(\vec{r})$. Electronegativity (χ) and hardness (η) are defined as (Pauling, 1960; Parr and Pearson, 1983):

$$\chi = -\mu = -\left(\frac{\partial E}{\partial N}\right)_{v(\vec{r})} \tag{2}$$

and η was defined by Parr and Pearson (1983) as the second order variation of E with respect to N:

$$\eta = \left(\frac{\partial^2 E}{\partial N^2}\right)_{v(\vec{r})} = \left(\frac{\partial \mu}{\partial N}\right)_{v(\vec{r})}$$
(3)

By applying finite difference approximation (Parr and Yang, 1989), μ and η can be expressed in terms of ionization potential (IP) and electron affinity (EA) at fixed $v(\vec{r})$ as:

$$\mu = -\frac{\left(IP + EA\right)}{2} \tag{4}$$

$$\eta = IP - EA \tag{5}$$

We have applied Koopmans' theorem for closed-shell molecules so as to avoid calculating E_{N} , E_{N+1} and E_{N-1} to get IP and EA and hence cutting down computational cost in the process:

$$\mu = \frac{E_{LUMO} + E_{HOMO}}{2} \tag{6}$$

$$\eta = E_{LUMO} - E_{HOMO} \tag{7}$$

 $E_{\rm HOMO}$ and $E_{\rm LUMO}$ being the energies of the highest occupied and lowest unoccupied molecular orbitals.

Electrophilicity index is known to build robust QSAR models with good correlation coefficients (R^2) in predicting biological activity (Roy et al., 2007) and the toxic behavior (Parthasarathi et al., 2004) of certain class of molecules.

COMPUTATIONAL DETAILS

Considering the type of elements present in the molecules, the geometry optimization is carried out at the Hartree-Fock (Hartree, 1928;1929; Fock, 1930) level along with a polarized split-valence 6-31G(d) basis set (Petersson et al., 1988) using the Gaussian 09 software (Frisch et al., 2014). Frequency analysis is performed at the same level to check whether there is any imaginary frequency to confirm that the optimized structures lie at the minima of their respective potential energy surfaces. Quantum chemical properties of the system like chemical potential (μ) and hardness (η) are computed by employing finite difference approximation and Koopmans' theorem, and the global electrophilicity descriptors ω and ω^2 are calculated using Parr's formula.

QSAR analysis for human African trypanosomiasis (HAT) healing activity of a set of pyridyl benzamides is performed against *T. brucei* using MLR technique in Origin 6.0 (Deschenes and David A. Vanden BoutUniversity of Texas, 2000). Sum of ranking differences (SRD) (Héberger, 2010;Kollár-Hunek and Héberger, 2013) is employed to compare the developed models using Comparison of Ranks by Random Numbers (CRRN) technique.

RESULTS AND DISCUSSION

Computer-aided molecular design and drug design utilize cheminformatics techniques like pattern recognition and data visualization to look for molecules with specific biological properties. Our *in silico* investigation allows prediction of biological activity in terms of electronic descriptors like electrophilicity (ω) and its square (ω^2) coupled with other descriptors like GATS8c, RDF40p and RDF55s within the domain of cheminformatics. Three-parameter QSAR analysis is performed in this dataset of 32 pyridyl benzamide derivatives taken from a study carried out by Masand *et al.* (2016) in the domain of cheminformatics. The dataset contains diverse substituents like methyl, fluoride, chloride, bromide, cyanide, methoxy, amino, *etc.* at different positions of the aromatic rings (Figure 1 and Table 1). The optimized geometries of a few of the compounds from the dataset are depicted





in Figure 2 as representative cases. The descriptors used in the aforementioned study were calculated using PaDEL 2.21 and were carefully sorted out using objective and subjective feature selection in QSARINS-Chem 2.2.1, taking into consideration various constitutional, 1D, 2D and 3D structural features. Out of the four models used in that work, we have considered models 1.1 and 1.2 for our study:

Sl. No.	R ₁	R ₂	Expt. pIC ₅₀	ω	ω^2	
1	Н	2-Me	5.5190	0.37	0.14	
2	4-CN	2-Me	5.6380	0.60 0.36		
3	4-Me	2-Me	5.6780	0.36	0.13	
4	4-Cl	2-Me	5.6780	0.44	0.19	
5	4-Br	2-Me	5.7450	0 0.44		
6	4-F	2-Me	6.2920	0.43	0.18	
7	4-C≡CPh	2-Me	5.7700	0.47	0.22	
8	4-C≡CCH ₂ -iPr	2-Me 5.9590		0.40	0.16	
9	4-Ph	2-Me	5.6990 0.40		0.16	
10	5-OMe	2-Me	5.4200 0.35		0.12	
11	6-CH=CH(CH ₂) ₂ CH ₃	2-Me	5.2220	0.31	0.10	
12	6-NH ₂	2-Me	5.3370	0.30	0.09	
13	Н	Н	5.0420	0.40	0.16	
14	Н	2-Et	5.5620	0.36	0.13	
15	Н	2-Me, 3-F	6.0510	0.41	0.17	
16	Н	3-F	5.1350	0.46	0.21	
17	Н	2-Me, 3-Cl	6.2010	0.40	0.16	
18	Н	2-Me, 3-Br	5.2440	0.41	0.16	
19	Н	2,3-diMe	5.4560	0.34	0.11	
20	Н	2-Me, 4-F	5.7210	0.38	0.15	
21	Н	2-Me, 4-Cl	5.9590	0.43	0.19	
22	Н	2-Me, 4-Br	5.9590	0.44	0.19	
23	Н	2,4-diMe	6.0090 0.36		0.13	
24	Н	3-F, 4-F	6.3870 0.47		0.22	
25	4-Me	3-F	6.0810	0.48	0.23	
26	4-Me	3-F, 4-F	6.0460	0.46	0.21	
27	4-C1	3-F	6.3280	0.53	0.28	
28	4-Cl	3-F, 4-F	6.2760 0.54		0.30	
29	4-F	3-F	7.0000	0.52	0.27	
30	4-F	3-F, 4-F	6.7210	0.53	0.29	
31	5-OMe	3-F	5.6380	0.43	0.19	
32	5-OMe	4-F	5.5850	0.39	0.15	

Table 1. Experimental plC₅₀, ω , ω^2 along with the substituents for all the compounds in the dataset are presented corresponding to the structural skeletons in Figure 1

Figure 2. Geometries of a few representative compounds from the dataset provided in Table 1 optimized at the HF/6-31G(d) level of theory



 $pIC_{50} = 4.8523(\pm 0.3676) - 0.9802 (\pm 0.4082)*GATS8c + 0.1240(\pm 0.0734)*RDF40p + 0.0552(\pm 0.0099)*RDF55s$ (1.1)

 $pIC_{50} = 6.4521(\pm 1.2282) - 3.1455 (\pm 2.3630)*E1s + 0.0545(\pm 0.0492)*RDF40m + 0.0489(\pm 0.0111)*RDF55s$ (1.2)

where, Geary autocorrelation of lag-8/weighted by atomic charges, Radial distribution function-040/ weighted by relative polarizabilities and Radial distribution function-055/weighted by relative I-state are abbreviated as GATS8c, RDF40p, and RDF55s respectively. E1 and RDF40m refer to 1st component accessibility directional WHIM index/weighted by relative I-state, and Radial distribution function - 040/weighted by relative mass.

As mentioned above, these descriptors were selected by screening a pool of descriptors generated using PaDEL 2.21. However, this descriptor calculator does not evaluate the electrophilicity index of a compound, which has been proven to be an effective quantum chemical descriptor in predicting toxicity (Parthasarathi et al., 2004) and biological activities (Roy et al., 2007). Thus, exploring the electronic environment of the chemical compounds in this regard makes sense. The electronic aspect of the compounds is taken care of in this study in the form of global electrophilicity index (ω) and its square (ω^2). These descriptors could have been used in addition to the ones used by Masand *et al.* to obtain QSAR models with better predictive ability, but in that case, over-fitting would have been a high possibility. To avoid that, we have developed separate models by replacing each of the above descriptors (in model 1.1 and 1.2) once with ω and again with ω^2 , followed by replacing two of them with both ω and ω^2 . All such possible combinations are done for models 1.1 and 1.2.

Now the division of set in our study is done by distributing the compounds in three sets (A, B and C, provided in Table 2) among which two of them are combined to form the training sets and the remaining is the test set. Thus, there are three such cases, *viz.*, *Case 1* contains set A as test and sets

	Set A	Set B	Set C
	29	30	24
	28	6	27
	17	25	15
	8	23	26
	21	22	7
Compound numbers ^a	9	20	5
	3	4	2
	14	32	31
	1	19	10
	11	18	12
	16	13	

Table 2. Division of 32 compounds taken for this study into three sets A, B and C. The numbers correspond to the compound numbers as presented in Table 1

^aas represented inTable 1

B & C as training, *Case 2* contains set B as test and A & C as training, and *Case 3* contains the only remaining combination *i.e.*, set C as test and A & B as training. The division of the dataset into 3 subsets is, however, not done randomly. Rather, the pIC_{50} values are used as a guide in this regard. The division is done in such a way to keep the average value of pIC_{50} of the compounds almost similar in each set. The QSAR models are built by simple multiple linear regression (MLR) technique where the coefficient of regression (\mathbb{R}^2) and the standard deviation (SD) bring out their predictive ability (Table 3). For relatively small datasets, models built on the entire undivided datasets are required to compare and assess the ones built after dividing the dataset into training and test sets (Masand et al., 2015).

Table 3. R ² and SD values obtained from MLR analysis on the undivided as well as three combinations of the divided da	taset
on 21 different QSAR models	

Model	Generalized Regression Equations	Undivided		Case 1		Case 2		Case 3	
No.		R ²	SD						
1	pIC ₅₀ =a+ b*GATS8c +c*RDF40p+ d*RDF55s	0.8284	0.1960	0.9182	0.1320	0.8216	0.1877	0.6971	0.2213
2	$pIC_{50} = a + b*GATS8c + c*RDF40p + d*\omega$	0.3743	0.3742	0.4839	0.1872	0.3764	0.1546	0.2351	0.4378
3	$pIC_{50} = a + b*GATS8c + c*\omega + d*RDF55s$	0.7599	0.2318	0.8325	0.1768	0.7829	0.2063	0.5114	0.2415
4	$pIC_{50} = a + b^*\omega + c^*RDF40p + d^*RDF55s$	0.7113	0.2542	0.7644	0.1983	0.7098	0.2283	0.4961	0.2529
5	$pIC_{50} = a + b*GATS8c + c*RDF40p + d*\omega^2$	0.3650	0.3770	0.4825	0.1827	0.3901	0.1447	0.2211	0.4098
6	$pIC_{50} = a + b*GATS8c + c*\omega^2 + d*RDF55s$	0.7592	0.2322	0.8323	0.1762	0.7853	0.2058	0.5817	0.2299
7	$pIC_{50} = a + b^*\omega^2 + c^*RDF40p + d^*RDF55s$	0.7068	0.2562	0.7620	0.1977	0.7101	0.2301	0.4645	0.2575
8	$pIC_{50} = a + b*GATS8c + c*\omega + d*\omega^2$	0.3285	0.3877	0.4334	0.1888	0.2506	0.1882	0.1725	0.4421
9	$pIC_{50} = a + b^*\omega + c^*RDF40p + d^*\omega^2$	0.3660	0.3767	0.4746	0.1781	0.2911	0.1800	0.1810	0.4594
10	$pIC_{50} = a + b^*\omega + c^*\omega^2 + d^*RDF55s$	0.6836	0.2661	0.7637	0.2014	0.6962	0.2212	0.5163	0.2364
11	$pIC_{50} = a + b*E1s + c*RDF40m + d*GATS6m$	0.3056	0.3942	0.3991	0.1983	0.3199	0.2404	0.1159	0.2055
12	$pIC_{50} = a + b*E1s + c*RDF40m + d*\omega$	0.3647	0.3771	0.4949	0.1839	0.3196	0.1784	0.2793	0.3939
13	$pIC_{50} = a + b*E1s + c*\omega + d*GATS6m$	0.4847	0.3396	0.5936	0.2275	0.5415	0.2138	0.3369	0.3171
14	$pIC_{50} = a + b*\omega + c*RDF40m + d*GATS6m$	0.4758	0.3425	0.5824	0.2177	0.5019	0.2127	0.3263	0.3134
15	$pIC_{50} = a + b*E1s + c*RDF40m + d*\omega^2$	0.3571	0.3793	0.4997	0.1769	0.3241	0.1706	0.2135	0.4273
16	$pIC_{50} = a + b*E1s + c*\omega 2 + d*GATS6m$	0.4763	0.3424	0.5839	0.2266	0.5473	0.2083	0.2648	0.3481
17	$pIC_{50} = a + b*\omega^2 + c*RDF40m + d*GATS6m$	0.4666	0.3455	0.5708	0.2181	0.5105	0.2030	0.2518	0.3441
18	$pIC_{50} = a + b*E1s + c*\omega + d*\omega^2$	0.3421	0.3421	0.4673	0.1865	0.2898	0.1777	0.1339	0.4681
19	$pIC_{50} = a + b^*\omega + c^*\omega^2 + d^*GATS6m$	0.4784	0.3417	0.5861	0.2301	0.4044	0.2680	0.1784	0.3648
20	$pIC_{50} = a + b^*\omega + c^*RDF40m + d^*\omega^2$	0.3583	0.3790	0.4822	0.1844	0.2185	0.2091	0.1454	0.4405
21	$pIC_{50} = a + b^*\omega + c^*\omega^2$	0.3272	0.3813	0.4540	0.1813	0.2922	0.1765	0.1288	0.4612

As we can see from the R² and SD values provided in Table 3, the models with the descriptor combinations 1, 3, 4, 6, 7 and 10 show good correlation coefficients (R²). For the rest of the models, the R² values lie below the threshold 0.6 which are statistically not relevant. Model 9 as reported in the study by Masand *et al.*, was based on multiple random divisions of the dataset. Here in this carefully sorted dataset splitting based on pIC₅₀ values does not produce good correlation coefficients. However, in certain cases, the R² value does reach close to 0.6 *e.g.*, for the combinations 13, 14, 16, and 17. The model containing the combination of GATS8c, RDF40p, and RDF55s as descriptors for the undivided dataset shows excellent R², and it is noted that removing RDF55s from the models drastically worsens the predictive ability of the models highlighting its importance. The efficiency with which a QSAR model predicts is reflected in the correlation coefficient between experimental and calculated activity (here pIC₅₀) of compounds that were not used for model development (*i.e.*, the test set). The correlation between the calculated and experimental pIC₅₀ values for model number 1 for all cases are depicted in Figure 3.



Figure 3. Plots of experimental versus calculated plC₅₀ along with their respective R² and SD values for the test sets of model 1

Further, we have employed the sum of ranking differences (SRDs) method as a model-comparison tool. In this method, relevant data need to be arranged in a matrix form where the rows and columns correspond to objects and variables, respectively. In this study, the objects refer to R² and SD values of the corresponding QSAR models (variables). The results of each model are ranked in increasing order, followed by taking a difference between the actual rank of the model results and that of the

reference or standard results (ideal). The sum of the absolute values of these differences is known as the SRD values which are used to compare the models. In the case where the ideal/reference ranking is unknown, the maximum/minimum/average value can be considered as ideal/golden standard depending upon the type of the objects. For example, in the present study, the maximum value of R^2 and minimum of SD are taken as reference/ideal in each row. This technique uses the concept of Occam's razor to evaluate the ranking: lower the value of SRD, better is the model. The graph (Figure 4) depicts the ranking of all the models used in this study, which also helps understand the extent of similarity between models from their relative position.

Figure 4. Evaluation of choice of descriptors using SRD with ties and subjected to the scaling between 0-100. Maximum R² and minimum SD values are used as golden standards. Scaled SRD values are plotted on the x-axis and left y-axis, the right y-axis shows the relative frequencies (black curve).



We have built 21 QSAR models on the undivided set and each of the three cases of divided sets as tabulated in Table 3. Thus our input matrix contains 21 columns of variables and 8 numbers of row (R² and SD for cases 1, 2, 3 and the undivided set). The program is so written such that if the nR (number of rows) is less than or equal to 8, it uses theoretical discrete distribution for the CRRN (Comparison of Ranks by Random Numbers) figure. From the graph, it can be easily inferred that the model with descriptors GATS8c, RDF40p and RDF55s having SRD value zero is the one with the best predictive ability, followed by models 3, 4, 7 and 6, 10 with SRD values of 6.25 and 12.50% respectively.

CONCLUSION

The current theoretical investigation of the QSAR model study is performed considering a dataset consisting of 32 pyridyl benzamide derivatives toward HAT activity using all possible combinations of

electronic factors (ω , ω^2) as easily computable descriptors in addition to the other descriptors against *T. brucei* as reported by Masand *et al.* The inclusion of electronic factors along with the screened descriptors on the developed models has a pretty good influence on predicting the biological activity of compounds. It is, however, noticed that when RDF55s is replaced with ω or ω^2 in model 1.1, the correlation coefficient (\mathbb{R}^2) value decreases by a significant amount suggesting its importance in describing the HAT activity. The robustness of the predicted models has been analyzed not only in terms of \mathbb{R}^2 (above the threshold value of 0.6) but also in the SRD-CRRN tool with ties to compare the efficacy of the models. CRRN integrated computational tool SRD helps us choose better models on the overall prediction of datasets consisting of dissimilar functionalities. The model constituted with descriptors GATS8c, RDF40p and RDF55s have the best predictive ability as reflected in the SRD graph (SRD value is zero) and models with number 3, 4, 7 and 6, 10 come next to it as the second and the third best predictive models having SRD values of 6.25 and 12.50% respectively.

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CONFLICT OF INTEREST

The authors declare that they have no conflict of interests regarding the publication of this article, financial, and/or otherwise.

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